

## Tetramethyl Orthosilicate (TMOS) detection by Advanced Ion Mobility Spectrometer - AIMS

The ion mobility spectrometry technique offers advantages like high sensitivity (ppb range), fast response (ms range), compact design, operation in atmospheric pressure and ability to separate the isomeric compounds. In this short report we demonstrate the sensitivity and fast response of IMS technique at low ppb level. As a case compound was chosen Tetramethyl Orthosilicate TMOS

The TMOS  $\text{Si}(\text{OCH}_3)_4$  of molar mass 152.25 g/mol, is frequently used in organic synthesis. The TMOS at low concentrations inhalation causes lung lesions, and at slightly higher concentrations eye contact with the vapor causes blindness. Due to this reasons is monitoring of TMOS in low concentration required especial in clean industrial hall where it is frequently used.

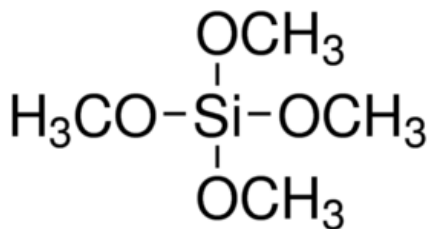


Fig.1. Tetramethyl Orthosilicate

In this short Laboratory Report we demonstrate the ability of Ion Mobility Spectrometer operated in sub-atmospheric pressure for continuous monitoring of TMOS at low ppb level.

### Experiment

The **Portable-Advanced Ion Mobility Spectrometer (PAIMS)** was used in this

experiment. The operating parameters of **PAIMS** are listed in Table 1.

Working pressure	<b>600 mbar</b>
Working temperature	<b>120 °C</b>
Drift Gas	<b>Zero Air</b>
Drift gas flow	<b>700 mL/min</b>
Drift field intensity	<b>570 V/cm</b>
Sample gas flow	<b>60 mL/min</b>
Polarity	<b>Positive</b>

Table1. **PAIMS** working parameters

The TMOS of analytical grade purity (Merck) was used in this experiment. The 1mL syringe of TMOS vapors diluted in ratio 1:20 with atmospheric air was used. The syringe was placed to syringe pump (Cronus) and interfaced via 2m long capillary of 0.15mm i.d. to **PAIMS** sample inlet. The long capillary with small i.d. was used in order to prevent diffusion. For calculation of concentration was used vapor pressure 12 mmHg (PubChem). The **PAIMS** operate in sub-atmospheric pressure and continuous sample sniffing was set to 60 mL/min. The sample inlet suck the atmospheric air, the vapors from the syringe was diluted to sample inlet flow by syringe pump.

Results and discussion

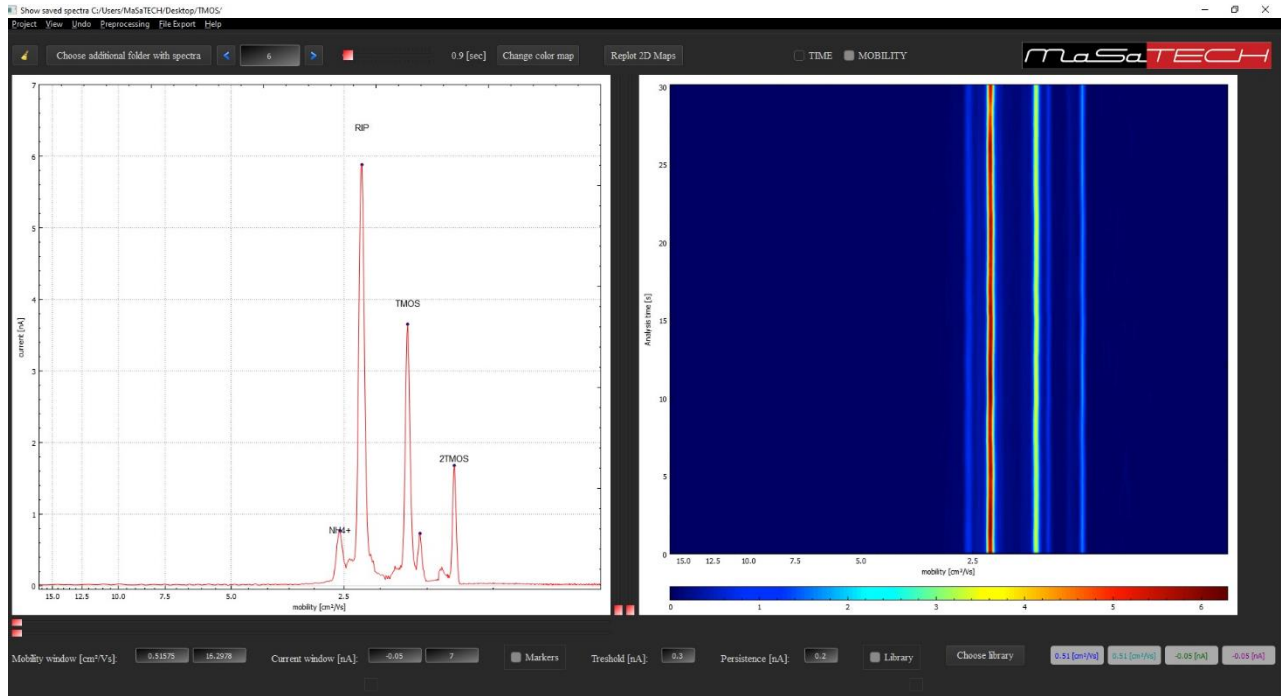


Figure 2. IMS response on 1.3 ppm of TMOS. Left IMS spectrum measured in reduced mobility mode, Right 2D map of 30s record time

The IMS response on 1.3 ppm of TMOS is shown in Figure 2. As we can see from this figure, there occurs the formation of two peaks: *TMOS* with reduced mobility  $1.69 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$  and *2TMOS* with reduced mobility  $1.27 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ . We suppose that there is going about the formation of **protonated monomer  $\text{H}^+ \cdot \text{TMOS}$**  (reduced mobility  $1.69 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ ) and **proton bound dimer  $\text{H}^+ \cdot (\text{TMOS})_2$**  (reduced mobility  $1.27 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ ). The 2D maps on the right side of Figure 2 show perfect stability of IMS response during the scan time 30s. During the experiment, also longer scan time was tested (20min) and only small deviation below 1% in peak intensity was observed.

The **Limit Of Detection (LOD)** for TMOS was measured directly and was determined **6.5 ppb**. Figure 3 shows the IMS response for syringe rate  $0.5 \mu\text{L}/\text{min}$  what represents **6.5 ppb**.

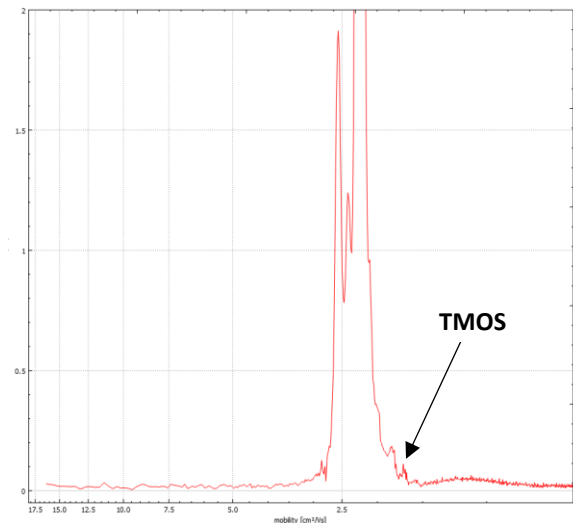


Figure 3. IMS response for 6.5 ppb of TMOS

The **MaSaTECH** software allows calculation of peak **volume**, peak **area**, averaged peak **area** along the monitoring time as well like peak **intensity** and averaged peak **intensity** along the monitoring time. The **PAIMS** results in good dynamic range from 6.5ppb to 3.9ppm as we can see from Figure 4.

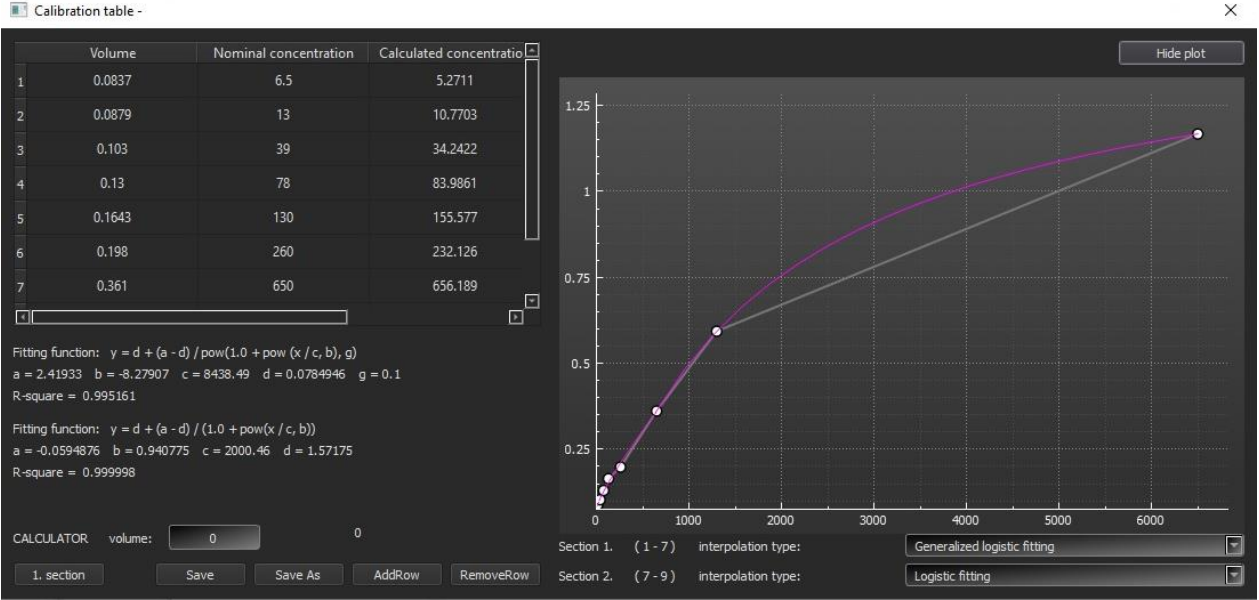


Figure 4. IMS response of TMOS from 6.5ppb to 3.9ppm

MaSaTECH software allow us to use several fitting functions like linear, exponential, logarithmic, logistic and generalized logistic. For TMOS the *generalized logistic in combination with logistic fitting* function seems to be appropriate with  $R^2=0.9951$ . The volume of peak area against the TMOS concentration was used for formation of calibration table.

**Fast Response**

The main advantage of linear Ion Mobility Spectrometers is related with fast response. The automatic peak derivation and unique measurements in reduced mobility mode allows our instruments fast peak detection and recognition. The online peak derivation giving also **real-time information** about intensity of target peak. The peak intensity can be used for immediate calculation of concentration for each chemical.

Peak Area (Arb.Unit)	Concentration ppb
0.0837	6.5
0.0879	13
0.103	39
0.13	78
0.1643	130
0.198	260
0.361	650
0.593	1300
1.167	3900

Table2 Calibration table for TMOS